

- J. Biol. Chem.* 236, 2530.
 Sotobayashi, H., Rosen, F., and Nichol, C. A. (1965),
Proc. Am. Assoc. Cancer Res. 6, 60.
 Usdin, E. (1959), *J. Biol. Chem.* 234, 2373.

- Werkheiser, W. C. (1962), *Proc. Am. Assoc. Cancer Res.*
 3, 371.
 Werkheiser, W. C. (1963), *Cancer Res.* 23, 1277.
 Zakrzewski, S. F. (1966), *J. Biol. Chem.* 241, 2957.

Coenzyme A Analogs. II. Enzymatic Conversion of D-Oxypantetheine 4'-Phosphate to Oxy-Coenzyme A*

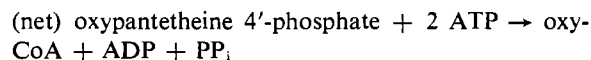
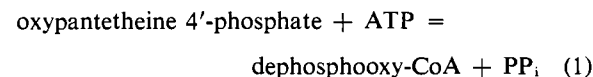
Charles J. Stewart and William J. Ball, Jr.†

ABSTRACT: The enzymatic conversion of D-oxypantetheine 4'-phosphate to oxy-coenzyme A was investigated with coenzyme A synthesizing enzymes isolated from beef liver. DEAE-cellulose column chromatography of the enzymatic incubation mixtures containing ATP-8-¹⁴C and D-oxypantetheine 4'-phosphate separated a radioactive component identical with synthetic

oxy-coenzyme A. Dephosphooxy-coenzyme A was enzymatically (a) cleaved by pyrophosphate to oxypantetheine 4'-phosphate and ATP and (b) phosphorylated by ATP to oxy-coenzyme A. Enzymatically prepared oxy-coenzyme A possessed the same inhibitory action as the synthetic material in the phosphotransacetylase reaction.

The oxygen analog of coenzyme A, oxy-coenzyme A, has been recently synthesized and shown to be a competitive inhibitor of CoA¹ in the phosphotransacetylase reaction (Stewart and Miller, 1965; Miller *et al.*, 1966). In order to determine if oxy-CoA can be prepared enzymatically, we investigated the ability of the crude CoA-synthesizing enzyme system, isolated from beef liver by the procedure of Hoagland and Novelli (1954), to convert D-oxypantetheine 4'-phosphate to oxy-CoA.

A net synthesis of oxy-CoA was achieved *via* reactions 1 and 2.



Experimental Procedures

Materials. D-Oxypantetheine 4'-phosphate, dephosphooxy-CoA, and oxy-CoA were prepared by the procedures of Miller *et al.* (1966). D-Pantetheine 4'-phosphate was prepared by the procedure of Moffatt and Khorana (1961). ATP-8-¹⁴C (lot no. 6501) was purchased from Schwarz BioResearch, Inc., and unlabeled ATP was purchased from Sigma Chemical Co. DEAE-cellulose (Selectacel standard type) was obtained from Carl Schleicher and Schuell. Phosphotransacetylase (lot no. 06155109) CoA, and acetyl phosphate were purchased from Boehringer Mannheim Corp. Lyophilized *Clostridium kluyveri* cells, as a source of crude transacetylase, and venom phosphodiesterase (*Crotalus adamantens*) were purchased from Worthington Biochemical Corp.

Methods. The Biuret reaction was used for protein determinations (Layne, 1957). Venom phosphodiesterase digestions of CoA and oxy-CoA were done by the previously described procedures (Moffatt and Khorana, 1961; Miller *et al.*, 1966). Phosphotransacetylase assays were performed by the procedure of Stadtman (1952) when preparing the beef liver enzymes, and as modified by Bergmeyer *et al.* (1963) for the inhibition studies. The ¹⁴C content of column fractions was determined by adding 0.1-ml aliquots to 14 ml of the aqueous counting medium of Bray (1960) and

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* From the Chemistry, Department, San Diego State College, San Diego, California 92115. Received July 28, 1966. This investigation was supported by Public Health Service Grant GM-07977 from the National Institute of General Medical Sciences. A preliminary report of a portion of this work has been published (Stewart *et al.*, 1966). Part I of this series is Miller *et al.* (1966).

† National Science Foundation Undergraduate Research Participant, 1964-1965.

¹ Abbreviations used: oxy-CoA, oxy-coenzyme A; desulfo-CoA, desulfocoenzyme A; dephospho-CoA, 3'-dephosphocoenzyme A; dephosphooxy-CoA, 3'-dephosphooxy-coenzyme A; AMP and ATP, adenosine mono- and triphosphates. The IUB systematic name for each of the following trivial enzyme names employed in this paper is: phosphotransacetylase, acetyl-CoA: orthophosphate acetyltransferase (EC 2.3.1.8); dephospho-CoA pyrophosphorylase, ATP:pantetheine 4'-phosphate adenylyltransferase (EC 2.7.7.3); dephospho-CoA kinase, ATP:dephospho-CoA 3'-phosphotransferase (EC 2.7.1.24).

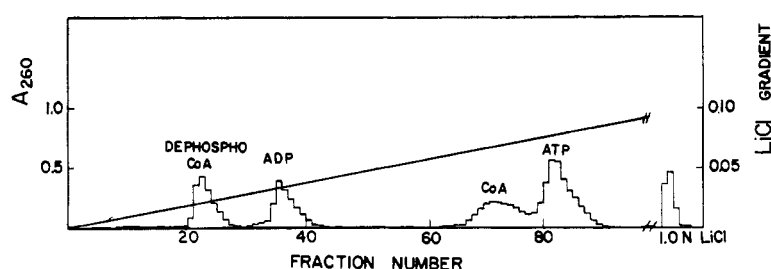


FIGURE 1: DEAE-cellulose column chromatography of pantetheine phosphate; ATP incubation mixture showing net enzymatic synthesis of CoA.

counted in a Packard Tri-Carb liquid scintillation counter. Paper chromatography was carried out by the descending technique on Whatman No. 1 paper, using ethyl alcohol-1 M ammonium acetate, pH 7.5 (7:3) solvent system. ADP and ATP were separated on DEAE-cellulose paper (Whatman Chromedia DE81) by descending elution with 10% aqueous formic acid. Adenine-containing compounds were located by their ability to absorb ultraviolet light (254 mμ). Phosphate-containing compounds were located by the method of Bandurski and Axelrod (1951). Table I lists the R_F values of the various compounds.

TABLE I: R_F Values of Compounds.

| Compound | System A ^a | System B ^b |
|------------------------------|-----------------------|-----------------------|
| CoA | 0.19 | |
| Dephospho-CoA | 0.48 | |
| Oxy-CoA | 0.16 | 0.02 |
| Dephosphooxy-CoA | 0.45 | 0.31 |
| Oxypantetheine 4'-phosphate | 0.57 | — |
| Adenosine 2'(3')-diphosphate | 0.04 | — |
| Venom digest of Oxy-CoA | 0.04, | — |
| | 0.57 | |
| ATP | 0.04 | 0.02 |
| ADP | 0.06 | 0.24 |
| AMP | 0.16 | 0.89 |
| Peak "X" | 0.15 | 0.34 |

^a Ethyl alcohol-1 M ammonium acetate, pH 7.5 (7:3). ^b Formic acid (10%)-DEAE-cellulose paper.

CoA-Synthesizing Enzymes and Incubations. The procedure of Hoagland and Novelli (1954) for the isolation from pig liver of a crude enzyme system containing dephospho-CoA pyrophosphorylase and dephospho-CoA kinase was successfully applied to beef liver. Incubations were carried out in a total volume of 2 ml at 38° for 1.5 hr and contained glycylglycine buffer, adjusted to pH 7.4 with Tris (80 μmoles), cysteine (20 μmoles), MgSO₄ (4 μmoles), ATP (5

μmoles), enzyme system (3 mg), and substrate. The substrates were: D-pantetheine 4'-phosphate (4 μmoles), D-oxypantetheine 4'-phosphate (4 μmoles), and dephosphooxy-CoA (2 μmoles). When dephosphooxy-CoA was incubated with 40 μmoles of inorganic pyrophosphate, ATP was omitted. The reactions were stopped by placing the incubation tubes in a boiling water bath for 1 min and removing the precipitated protein by centrifugation.

DEAE-Cellulose Chromatography. The centrifuged incubation mixtures were diluted to a total volume of 25 ml and applied to the top of a 1.3 × 50 cm DEAE-cellulose (chloride form) column. The column was washed with 0.003 N HCl until the effluent was free of ultraviolet-absorbing material. The adsorbed compounds were eluted by application of an acidic lithium chloride linear gradient. The reservoir contained 350 ml of 0.12 N LiCl in 0.003 N HCl and the mixing vessel contained 350 ml of 0.003 N HCl. Fractions (5 ml) were collected at a flow rate of 0.5 ml/min. Each ultraviolet-absorbing peak was pooled. After adjusting the pH to 4.5 with dilute solutions of LiOH, the peaks were evaporated to dryness *in vacuo*. Lithium chloride was removed from the solid white residues by repeated extractions with small volumes of methyl alcohol-acetone (1:15). After drying over P₂O₅ *in vacuo* at room temperature overnight the residual material was dissolved in 1 ml of water and characterized by paper chromatography.

The concentration of each peak was calculated from the optical density units at 260 mμ, assuming an extinction coefficient of 15×10^3 for the adenosine moiety.

Results

The validity of our experimental approach was tested by incubating pantetheine 4'-phosphate with ATP. Figure 1 clearly demonstrates the synthesis of CoA, 8.4 ODU, and its effective separation from the products of the enzymatic reaction mixture by DEAE-cellulose column chromatography. In addition to CoA, dephospho-CoA (8.7 ODU) and ADP (8.4 ODU) were produced in the reaction. The 1 N LiCl elution peak contains 5.2 ODU presumably, the mixed disulfide forms of CoA.

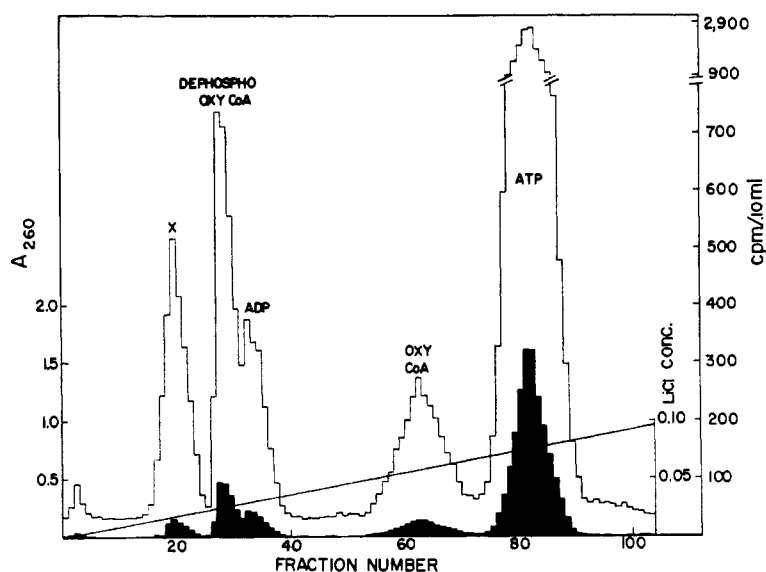


FIGURE 2: DEAE-cellulose column chromatography of oxypantetheine phosphate; ATP incubation mixture showing net enzymatic synthesis of oxy-CoA. The shaded area is the A_{260} elution profile and the upper curve is the elution pattern of radioactivity. Peak X is unidentified.

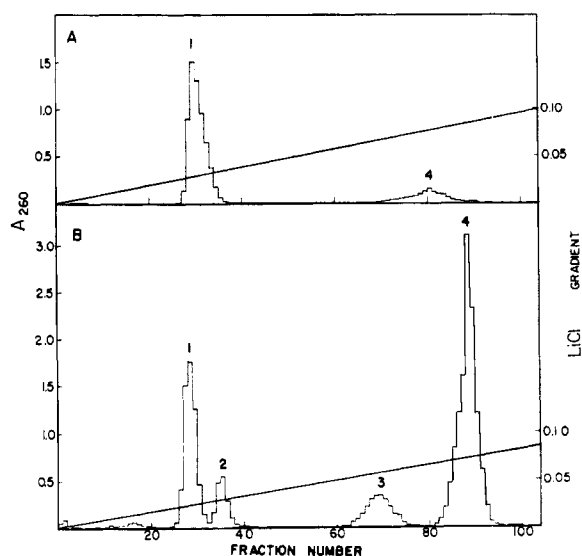


FIGURE 3: DEAE-cellulose column chromatography of dephosphooxy-CoA incubation mixtures. (A) Dephosphooxy-CoA incubated with inorganic pyrophosphate and (B) dephosphooxy-CoA incubated with ATP. Peak 1, dephosphooxy-CoA; peak 2, ADP; peak 3, oxy-CoA; peak 4, ATP.

The elution pattern of the incubation mixture containing oxypantetheine 4'-phosphate and ATP-8- ^{14}C is shown in Figure 2. The peaks containing the radioactive label are directly superimposed upon the ultra-violet-absorbing peaks. The yield of radioactive oxy-CoA was 6.5 ODU. The unidentified peak, "X," contained 3.6 ODU. The presence of label, a positive

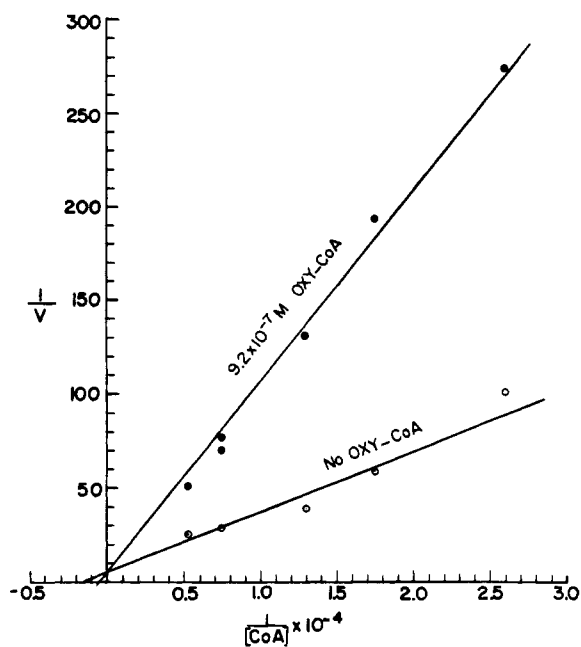


FIGURE 4: Double-reciprocal plot showing competitive inhibition of the phosphotransacetylase reaction by oxy-CoA isolated from peak 3, Figure 3B. The ordinate is the number of seconds required to observe an $\text{OD}_{233\text{m}\mu}$ change of 0.1, after addition of enzyme. The concentration of oxy-CoA is estimated from its absorbency at 260 $\text{m}\mu$.

reaction to the phosphate spray reagent, the elution position of "X," and its R_F in system A (*cf.* Table I) implicated AMP. However its R_F on DEAE paper

was not similar to AMP. Paucity of this material hindered further attempts at identification.

The ability of the crude enzyme preparation, containing dephospho-CoA pyrophosphorylase and dephospho-CoA kinase, to cope with the dephosphooxy-CoA is confirmed in Figure 3. The reversibility of the dephospho-CoA pyrophosphorylase reaction is indicated in Figure 3A. Incubation of 2 μ moles of dephosphooxy-CoA with 40 μ moles of sodium pyrophosphate produced 5.8 ODU (0.38 μ mole) of ATP. Dephospho-CoA kinase readily catalyzes the phosphorylation of dephosphooxy-CoA by ATP (Figure 3B). The oxy-CoA peak contained 11.2 ODU (0.75 μ mole) and was isolated in a 38% yield based upon the substrate concentration of 2 μ moles.

The identity of oxy-CoA isolated from peak 3, Figure 3B, was verified by inhibition studies with the phosphotransacetylase reaction in addition to routine paper chromatography and venom phosphodiesterase digestion. The K_i , 4.2×10^{-7} (Figure 4), is consistent with the value, 3.5×10^{-7} , reported for the crystalline synthetic compound (Miller *et al.*, 1966).

Discussion

This study demonstrates the enzymatic conversion of D-oxypantetheine 4'-phosphate to oxy-CoA. The results indicate that the thiol group is not intimately involved in the *in vitro* conversion of pantetheine 4'-phosphate to CoA by the beef liver system. An ancillary result of this study is the demonstration of a feasible enzymatic route for the synthesis of oxy-CoA or CoA labeled with ^{14}C in the adenosine moiety.

These results also may explain the effect oxypantetheine has on *Lactobacillus helveticus*. Oxypantetheine

is a competitive growth inhibitor of this pentetheine-requiring microorganism (Stewart *et al.*, 1955; Pierpoint *et al.*, 1955). It is probable that oxypentetheine inhibits the growth of *L. helveticus* by an *in vivo* synthesis of oxy-CoA. While not proving this proposition, this study supports it, especially if oxy-CoA *in vivo* has as broad an enzyme spectrum of CoA antagonism as Chase *et al.* (1966) have reported for desulfo-CoA *in vitro*.

References

- Bandurski, R. A., and Axelrod, B. (1951), *J. Biol. Chem.* 193, 405.
- Bergmeyer, H. U., Holz, G., Klotzsch, H., and Lang, G. (1963), *Biochem. Z.* 338, 114.
- Bray, G. A. (1960), *Anal. Biochem.* 1, 279.
- Chase, J. F. A., Middleton, B., and Tubbs, P. K. (1966), *Biochem. Biophys. Res. Commun.* 23, 208.
- Hoagland, M. B., and Novelli, G. D. (1954), *J. Biol. Chem.* 207, 757.
- Layne, E. (1957), *Methods Enzymol.* 3, 450.
- Miller, T. L., Rowley, G. L., and Stewart, C. J. (1966), *J. Am. Chem. Soc.* 88, 2299.
- Moffatt, J. G., and Khorana, H. G. (1961), *J. Am. Chem. Soc.* 83, 633.
- Pierpoint, W. S., Hughes, D. E., Baddiley, J., and Mathias, A. P. (1955), *Biochem. J.* 61, 190.
- Stadtman, E. R. (1952), *J. Biol. Chem.* 196, 527.
- Stewart, C. J., Cheldelin, V. H., and King, T. E. (1955), *J. Biol. Chem.* 215, 319.
- Stewart, C. J., and Miller, T. L. (1965), *Biochem. Biophys. Res. Commun.* 20, 433.
- Stewart, C. J., Miller, T. L., and Ball, W. J., Jr. (1966), *Federation Proc.* 25, 217.